Effects of heat and freeze on isolated erythrocyte submembrane skeletons

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Abstract. In this study we heated insoluble residues, obtained after Triton-X-100 (0.1 v/v%) extraction of erythrocyte ghost membranes (EGMs). Specific heat capacity, electric capacitance and resistance, and optical transmittance (280 nm) sustained sharp changes at 49°C ($T_A$) and 66°C ($T_C$), the known denaturation temperatures of spectrin and band 3, respectively. The change at $T_A$ was selectively inhibited by diamide (1 mM) and taurine mustard (1 mM) while its inducing temperature was selectively decreased by formamide in full concert with the assumed involvement of spectrin denaturation. In the residues of EGMs, pretreated with 4,4’-diisothiocyanato stilbene-2,2’-disulfonic acid (DIDS), the change at $T_C$ was shifted from 66 to 78°C which indicated the involvement of band 3 denaturation. The freeze and rapid thaw of EGM residues resulted in a strong reduction of cooperativity of band 3 denaturation while the slow thaw completely eliminated the peak of this denaturation. These effects of freeze-thaw were prevented in residues obtained from DIDS-treated EGMs. The freeze-thaw of residues slightly affected spectrin denaturation at 49°C although an additional denaturation appeared at 55°C. The results indicate preserved molecular structure and dynamics of the membrane skeleton in Triton-X-100 extracts of EGMs. The freeze-thaw inflicted strong damage on band 3 and spectrin-actin skeleton of EGM extracts which is relevant to cryobiology, cryosurgery and cryopreservation of cells.

Key words: Triton-X-100 shells — Spectrin — Band 3 — Freeze-thaw — Cryodamage

Abbreviations: Diamide, diazenedicarboxylic acid bis(N,N-dimethylamide); DIDS, 4,4’-diisothiocyanato stilbene-2,2’-disulfonic acid; DSC, differential scanning microcalorimetry; EGMs, erythrocyte ghost membranes; taurine mustard (taumustine), 2-[bis(2-chloroethyl)amino]ethanesulfonic acid; Triton-X-100, polyethylene glycol p-(1,1,3,3-tetramethylbutyl)-phenyl ether.

Introduction

The unique mechanical properties (deformability, elasticity and stability) of human erythrocytes are mainly due to the erythrocyte plasma membrane and the association of its membrane skeleton of peripheral proteins with its two major integral proteins, glycoporphin and band 3 (AE1, the anion exchanger) (Waugh and Agre 1988; Low et al. 1991). The membrane skeleton includes the third major protein of erythrocyte plasma membrane, spectrin, and a set of minor proteins such as actin, ankyrin, band 4.1, band 4.2, etc (Mohandas and Gallagher 2008). While the deformability of erythrocyte plasma membrane depends primarily on the state of spectrin-actin-glycophorin linkage, its mechanical stability chiefly depends on the linkage of spectrin via ankyrin to the tetrameric forms of band 3 (Van Dort et al. 1998; Beutler et al. 2000; Alberts et al. 2008) and on the interaction of spectrin with the negatively charged lipids within the internal monolayer of lipid bilayer (Hendrich et
al. 1991; Michalak et al. 1994; Zwaal and Schroit 1997). The remaining portion of band 3 is in dimeric form and includes copies which freely diffuse within the lipid bilayer or are linked to glycoporphin (Tanner, 1993; Reithmeier et al. 1996).

Erythrocyte shape stability and transformation are subject to a complex mechanism based on the minimum of the free energy of erythrocyte plasma membrane (Iglč et al. 1997; Mukhopadhyay et al. 2002). The latter includes the elastic energy of membrane skeleton and the bending energy of lipid bilayer as well as the free energy of embedded integral proteins (Kralj-Iglč et al. 1996). This mechanism involves also the ratio of external to internal monolayer area which could be affected by the conformation of major band 3 protein (Gimsa and Ried 1995). Results of Hianik et al. (2000) demonstrated that the role of the membrane skeleton probably involves maintaining a higher compressibility of erythrocyte plasma membranes.

The plasma membranes of many nonerythroid cells of mammals and birds also contain similar anion exchanger protein (Alper 1991) and submembraneous spectrin-actin network (Mangeat 1988). In humans, impairment of erythrocyte deformability and elasticity has been correlated with many pathologic conditions, such as myocardial infarction, diabetes mellitus, essential hypertension, hereditary spherocytosis, sickle cell anemia, and malaria (Chien 1987; Simchon et al. 1987; Simchon et al. 1991; Michalak et al. 1994; Zwaal and Schroit 1997). The cytoplasmic domain of band 3 protein denatures at 67°C (C peak) (Snow 1978). A more simplified model of EGMs is frequently obtained by extracting EGMs with mild, polar detergents, typically Triton-X-100. This procedure releases the erythrocyte plasma membrane skeletons as insoluble residues, substantially delipidated and devoid of the most of membrane integral proteins (Sharma and Gokhale 2011). These, so called Triton-X-100 shells (residues) contain chiefly spectrin, actin and 9%, respectively, of their total residual protein (Lux et al. 1976). The cytoplasmic domain of band 3 protein comprising about 75%, 5% and 9%, respectively, of their total residual protein (Lux et al. 1976).

Recent studies (Ivanov et al. 2012; Ivanov and Paarvanova 2016) have noted changes in the dielectric properties (complex impedance, Z*; complex capacitance, C*; dielectric loss curve) of human erythrocytes and their isolated impermeable EGMs and Triton-X-100 shells, induced at the spectrin denaturation temperature (49.5°C). Based on the temperature dependence of the loss curve, the spectrin skeleton of EGMs was considered dielectrically active only at native state. Hence, the respective changes in Z* and C* at 49.5°C were considered as contribution of spectrin skeleton to the dielectric properties of native EGMs. Based on the strong frequency dependences of the changes in Z* and C* at 49.5°C, the methods of dielectric spectroscopy (Klösgen et al. 2011) revealed two dielectric relaxations on the membrane skeleton of EGMs. The first relaxation depended on the availability of linkages between the integral proteins and spectrin and was detected at such frequencies whereat the lipid bilayer did not allow penetration of the field into cytosole. It was present on erythrocytes and impermeable EGMs and not on Triton-X-100 shells. It was explained as a direct piezoeffect on the flexible spectrin filaments with a mechanical force originating from the frequency-dependent charging of lipid bilayer. The second relaxation was detected at higher frequencies allowing direct interaction of alternating field with the spectrin dipoles, presumably those formed on the triple-helical repeat units of spectrin monomers.

The present study is aimed at the exporation of Triton-X-100 shells of EGMs with emphasis on dielectric changes related to the thermal denaturations of their major proteins, spectrin and band 3. These denaturations were detected by ultraviolet spectrophotometry and DSC. Thermal dielectros copy detected two major dielectric relaxations of above mentioned second type, coupled to these denaturations. A study of these thermally-induced dielectric changes showed that freeze-thaw of shells altered the band 3 and spectrin, and the alternation of the former was more pronounced than that of the latter. The rapid thaw of frozen shells strongly modified the band 3 denaturation while the slow thaw resulted in its complete elimination.

Materials and Methods

Materials

DIDS (4,4′-diiso-thiocyanato stilbene-2,2′-disulfonic acid), MgCl₂, NaCl, phosphate buffer, diamide (diazenedicarboxylic acid bis(N,N-dimethylamide)), taurine mustard (2-[bis(2-chloroethyl)amino]ethanesulfonic acid), Triton-X-100 (polyethylene glycol p-(1,1,3,3-tetramethylbutyl)-phenyl ether), EDTA (ethylenediaminetetraacetic acid) and formamide were purchased from Sigma Chemicals Co, St. Louis, MO, USA.
Isolation of erythrocytes

Human blood samples were obtained from healthy donors (donor centres at Stara Zagora, Bulgaria and Charité, Berlin, Germany). Erythrocytes were isolated by centrifugation (500 × g, 5 min) and three times washed in excess volume of cold 150 mM NaCl saline prior usage.

Thermal stabilization of band 3 with DIDS

DIDS is a membrane impermeable bifunctional covalent amino reagent, specifically binding and cross-linking band 3 monomer (Cabantchik and Greger 1992). Washed erythrocytes were suspended at hematocrit of 0.10 in 150 mM NaCl saline, containing 5 mM phosphate buffer, pH 7.8, and 50 mM DIDS at dark and at room temperature for 15 min. The DIDS-treated cells were isolated, twice washed in excess volume of cold 150 mM NaCl saline to remove non-reacted DIDS. Under these conditions, more than 95% of the DIDS resides on band 3 (Jennings and Passow 1979) resulting in strong thermal stabilisation of this protein and shifting of its denaturation temperature from 67°C to 80°C (Snow et al. 1978).

Isolation of EGMs

One volume of cold, dense suspension of washed erythrocytes (intact or DIDS-treated), hematocrit 0.60, was rapidly and vigorously diluted in 15 volumes of 1°C-cold hypotonic solution, containing 4 mM MgCl₂ and 5 mM phosphate buffer, pH 7.8 and left at 4°C for 5 min. The obtained leaky EGMs, used for Triton-X-100 extraction, were isolated by centrifugation (4000 × g, 12 min) and additionally washed in the indicated hypotonic solution (Dodge et al. 1963).

Preparation of Triton-X-100 shells of erythrocyte ghost membranes

One ml of cold, leaky EGMs were vigourously mixed with equal volume of cold, washing solution (10 mM NaCl saline, 4 mM of MgCl₂, 5 mM phosphate buffer, pH 7.8), that contained 0.2% (w/v) Triton-X-100. When specially indicated, the EGMs were extracted at concentrations of Triton-X-100 higher than 0.1%. The contact of membranes with detergent immediately turned their opaque suspension into a transparent medium. As the Triton-X-100 itself is a chelator of Mg²⁺, higher concentration (4 mM) of these ions were used. Phosphate buffer was preferred to Tris as Tris solubilises and disrupts the Triton-X-100 shells. The extraction of EGMs lasted 30 min at 4°C. The obtained solution of crude, insoluble residues was diluted by 3 volumes of the indicated cold, washing medium and centrifuged (8000 × g, 12 min) to sediment the residues (Yu et al. 1973). The packed white paste of crude residues (about 0.2 ml) was immediately studied or, prior to study, additionally washed one, two or three times in excess volume of the washing medium. The term Triton-X-100 shells (erythrocyte membrane skeletons, spectrin-actin skeletons) further means three-step washed shells, if not specifically indicated.

During the Triton-X-100 extraction of EGMs the residual protease activity could disrupt the proteins of membrane skeletons, as shown by Ciana et al. (2005, 2011). This protease activity originates from the white blood cell contamination of erythrocyte preparation. To reduce it at the stage of erythrocyte isolation, the buffy coat containing the majority of white blood cells, was carefully removed and the erythrocytes were thrice washed in excess volume of cold 150 mM NaCl saline. The usage of EGMs, isolated after the hypotonic lysis of erythrocytes, additionally reduced protease activity.

Differential scanning calorimetry of Triton-X-100 shells

DSC measurements were performed using the DSC unit of a MSC Microcalorimeter, MicroCal, MA, USA. The measuring cell contained 1.442 ml Triton-X-100 shells, diluted in the washing medium at a volume ratio of 1:0.3. The reference cell was filled with the washing medium. Scanning was carried out with a heating rate of 0.7°C/min in the range from 20 to 90°C. Only two portions of each sample were scanned if there was no substantial difference between the obtained thermograms.

Spectrophotometric determination of thermal denaturations in Triton-X-100 shells

2.6 ml of a solution containing 10 mM NaCl, 4 mM Mg²⁺ and 5 mM phosphate buffer, pH 7.4, was placed in a quartz glass cuvette and the optical transmittance at 280 nm (T₂₈₀) was set at 100%. Thereafter 40 μl of the Triton-X-100 shell paste was injected in the solution reducing T₂₈₀ to about 40%. The optical transmittance was measured by a UV spectrophotometer (Milton Roy Spectronic 21D, USA), equipped with an electric heater of the measuring cuvette. Thermal denaturations were recorded heating the cuvette at a rate of 2°C/min (Poklar et al. 1999). The temperature t (°C) of the solution was measured by an electronic thermometer (accuracy 0.1°C). The analog output signals of the spectrophotometer and the thermometer were both fed to a computer through a two-channel analog-to-digital converter.

The thermal denaturation of a test protein (fibrinogen) caused a sharp decrease in the optical transmittance as a result of the unfolding of its polypeptide chain and the change (typically decrease) in its molar volume. Spectrophotometric profile of the denaturation of fibrinogen (0.03 wt.%), dissolved in 140 mM NaCl and 5 mM phosphate buffer, pH 7.4
is shown in Figure 1. The obtained $T_{280}/t^\circ$C profile demonstrated the denaturation as a sigmoid decrease with midpoint at the temperature of denaturation, $T_d$. The derivative curve, $dT_{280}/dt$, in Figure 1 corresponds to the rate of denaturation and specifies the top temperature of denaturation, $T_d$, and the half-width, $\Delta T_{1/2}$, of the denaturation peak.

**Thermal dielectroscopy of Triton X-100 shells**

The sample cuvette (a cylindric glass tube with length = 120 mm, outer diameter = 4 mm and wall width = 0.5 mm) contained two platinum wire electrodes mounted at a distance of about 4 mm from each other. Prior to heating, 70 μl sample of the tested paste of Triton X-100 shells was placed in the cuvette and deaerated at room temperature under vacuum for 5 min. The cuvette was put into a hole, drilled into an aluminum block. The block was heated with constant heating rate. During heating an alternating electric voltage of 80 mV was imposed between the electrodes and the complex impedance, $Z^* = Z_{re} + jZ_{im}$, and complex capacitance, $C^* = C_{re} + jC_{im}$, of the sample were continuously measured and separated into their real ($Z_{re}$ and $C_{re}$) and imaginary ($Z_{im}$ and $C_{im}$) parts (Ivanov 2010). Here, $j$ is the imaginary unit, $j = (-1)^{0.5}$. $Z^*$ and $C^*$ were measured in a sweep frequency mode at 12 frequencies between 0.05 and 10 MHz, scanned serially with integration time of 0.5 s. The duration of each scan was less than 10 s. The core instrument was a Solartron 1260 Impedance Analyzer (Schlumberger Instruments, Hampshire, England) interfaced to Toshiba PC using the Miniscan software.

Platinum electrodes, low sample conductance (about 40 μS), low electrode voltage (80 mV) and frequencies above 50 kHz were used in order to decrease electrode polarisation. The Impedance Analyzer removes the noise and harmonic distortions and has 0.1% accuracy and 0.001 dB resolution.

**Results**

Micrographs, taken with confocal microscope, showed the crude Triton-X-100 shells as pale mostly cup-shaped bodies with a diameter close to that of native erythrocytes (Fig. 2). Each shell was covered with a number of small vesicles adhering to it. The maximum diameter of these vesicles was 0.30 μm. Due to the low extracting concentration of the detergent, these vesicles could contain residual amounts of EGM lipids and integral proteins, mostly band 3.

**Thermal denaturation of spectrin and band 3 of Triton-X-100 residues**

Upon heating thrice washed Triton-X-100 shells the optical transmittance at 280 nm indicated two sigmoid changes with transition mid-temperatures at $T_A$ (49°C) and $T_C$ (66°C) (Fig. 3, lower curve). The temperature derivative of optical transmittance (Fig. 3, upper curve) exhibited the indicated changes as narrow peaks (A and C), centred at the above mentioned $T_A$ and $T_C$ temperatures. The narrow half-widths and high top temperatures of these peaks, compared to the denaturation peak of the test protein (Fig. 1), indicate involvement of protein denaturation in both events.

DSC scan of thrice washed Triton-X-100 shells exhibited two peaks of heat absorptions, centered at the same inducing temperatures, $T_A$ (49°C) and $T_C$ (66°C) (Fig. 4, curve 1). At the same inducing temperatures similar peaks have been calorimetrically obtained with samples of isolated EGMs and attributed to the heat denaturations of spectrin (the...
calorimetric A peak) and the integral domain of band 3 (the calorimetric C peak), respectively (Brandts et al. 1977; Ivanov et al. 2007). Hence, it could be assumed that the peaks at the $T_A$ and $T_C$ top temperatures on the spectrophotometric (Fig. 3) and DSC (Fig. 4) thermograms of Triton-X-100 shells corresponded to the heat denaturations of spectrin and band 3, respectively.

To additionally inspect these two denaturations, we subjected Triton-X-100 shell samples to continuous heating and measured the complex impedance and complex capacitance at various frequencies. At temperatures above 45°C, two irreversible sigmoid changes were registered on the temperature profiles (thermograms) of the real and imaginary parts of these parameters, reflecting thermal damage to shells. For example, the changes on the thermogram of real capacitance, $C_{re}$, measured at 1 MHz, are shown in Fig. 5. The amplitudes of the two changes, $\Delta C_A$ and $\Delta C_C$ respectively, demonstrated remarkable frequency dependence (not shown). In order better to determine the inducing temperatures of these changes the temperature derivative of the sample capacitance ($dC_{re}/dt$) was also obtained (Fig. 5). The inducing temperatures of these changes exactly coincided with above indicated denaturation temperatures, $T_A$ and $T_C$, respectively.

With once washed Triton-X-100 shells, the above indicated two sigmoid changes on the capacitance and resistance thermograms were centred at lower temperatures, 46°C and 55°C, respectively (not shown). Increasing the number of washings to three, the mid-temperatures of these changes shifted upwards to the values, indicated in Fig. 5 and any additional wash did not produced effect on these temperatures. This result indicates that the residual amounts of Triton-X-100, remaining in the shells, reduced the above mentioned denaturation temperatures. The complete removal of Triton-X-100 resulted in a less pronounced temperature shift (3°C, from 46 to 49°C) of the first change and much more significant shift (11°C, from 55 to 66°C) of the second change. Thus, the effect of Triton-X-100 on the denaturation temperatures of spectrin and band 3 correlated the different ability of Triton-X-100 to bind and destabilize the peripheral and integral proteins of EGM, respectively.

The membrane skeletons of EGMs were also obtained after extraction of leaky EGMs with Triton-X-100 at the concentrations of detergent higher than 0.1%. Typically 0.1% (v/v) Triton-X-100 is sufficient to lyse erythrocytes, and even up to 0.4% concentrations will usually not harm most enzymes which are isolated. The sediment of insoluble residues, obtained at higher Triton-X-100 concentrations, had smaller volume. With increasing concentration of detergent from 0.2 to 0.4% the $\Delta C_A$ and $\Delta C_C$ amplitudes, as defined in Fig. 5, were reduced and at 0.8% these amplitudes became undetectably small (not shown). This result possibly indicates that EGM proteins became denatured by the detergent at the stage of extraction due to the high detergent concentration.

DIDS is a membrane-impermeable amino reagent, covalent inhibitor of band 3 of native erythrocytes. At low (<50 μM) concentrations in outside medium it specifically binds band 3, increasing its denaturation temperature step-wisely by 13°C (Snow et al. 1978). Using Triton-X-100 shells,
isolated from DIDS-treated erythrocytes, similar step-wise shift of $T_C$ from 66°C to 78°C was registered on their DSC thermogram (Fig. 4, curve 2) and on their spectrophotometric, capacitance and resistance thermograms (not shown). At the same time, DIDS had no affect on the inducing temperature ($T_A$) of the first change. The same step-wise shift of $T_C$ was obtained adding DIDS (final concentration 50 μM) to Triton X-shells, isolated from intact EGMs, during the first wash of shells. In conclusion, this specific threshold effect of DIDS on thermal stability of Triton-X-100 shells, strongly supports the involvement of band 3 denaturation in the change at $T_C$.

Diamide is a bivalent sulfhydryl reagent while taurine mustard is a dialkylating agent both specifically cross-linking spectrin dimers, when applied at low (1 mM) concentration on native erythrocytes (Fischer et al. 1978; Wildenauer et al. 1980). While it is not bound itself, diamide mediates the formation of disulfide bonds between spatially adjacent sulfhydryl groups primarily in spectrin. Both inter- and intra-molecular disulfide bonds are formed: about one per 30 spectrin dimers of the former type and one per three spectrin dimers of the latter type. In this study, we exploited the specific action of diamide and taurine mustard toward spectrin, adding these reagents to thrice washed Triton-X-100 shells at the final concentration of 1 mM. Both reagents inhibited about 3-fold stronger the amplitude of the sigmoid change at $T_A$ compared to that at $T_C$ (not shown), whereas the values of $T_A$ and $T_C$ temperatures were not affected. This result is in line with the assumed involvement of spectrin in the denaturation of Triton-X-100 shells at $T_A$. In addition, this result indicated that specific cross-linking and resulting immobilisation of intramolecular segments of spectrin reduced the dielectric polarisation of Triton-X-100 shells.

Polar organic solvents are heavily used in various biomedical applications. For example, formamide lowers the melting temperatures of deoxyribonucleic acids linearly by 2.4–2.9°C/mol of formamide (Blake and Delcourt 1996). This effect is used to accelerate the hybridization of deoxyribonucleic acids. Applied on native erythrocytes and impermeable EGMs, the great majority of polar organic solvents destabilised much stronger band 3 than spectrin (Ivanov 2001; Paarvanova et al. 2012). As an exception, formamide displayed a specific ability to destabilise spectrin, decreasing its denaturation temperature at about 3 time stronger rate compared to that of band 3. In this study, formamide demonstrated even more its specific effect on the proteins of Triton-X-100 shells. At concentrations less than 7% (v/v) it reduced linearly the $T_A$ and $T_C$. The former, however, was reduced at about a 10-fold higher rate compared to the latter (not shown). This finding is in line with the assumed participation of spectrin in the denaturation at $T_A$.

Added to the Triton-X-100 residues prior to heating, EDTA strongly reduced the amplitudes, $\Delta C_A$ and $\Delta C_C$, and decreased the inducing temperatures, $T_A$ and $T_C$, on the capacitance and resistance thermograms (not shown). For example, 5 mM EDTA inhibited about three times the capacitance changes $\Delta C_A$ and $\Delta C_C$ (as defined in Fig. 3), and decreased the $T_A$ and $T_C$ by 4°C and 11°C, respectively. This result is in line with the known ability of EDTA to disrupt the structure and solubilize the proteins of EGMs (McMillan and Luftig 1973).

**Dielectric relaxations on spectrin and band 3 of Triton-X-100 residues**

As noted above the amplitudes of detected changes at $T_A$ and $T_C$ on the complex impedance and capacitance thermograms of Triton-X-100 residues displayed strong frequency dependences. These frequency effects reflect the contribution of spectrin and band 3 to the dielectric polarisation of Triton-X-100 residues as the amplitudes of both dielectric changes at $T_A$ and $T_C$ were reduced to zero after the thermal denaturation of respective proteins. To inspect both frequency dependences in more detail, dielectroscopy methods were further applied. They allow get insight into the molecular mobility of electric dipoles associated to proteins involved in respective dielectric relaxations. In contrast to the measurement of sample impedance, the measurement of
The sample capacitance was impeded at frequencies, lower than 100 kHz due to the electrode polarisation. Thermograms of complex impedance, \( Z^* = Z_{re} + Z_{im} \), were therefore used to investigate the frequency dependences of the amplitudes of the dielectric changes at \( T_A \) and \( T_C \).

The amplitudes of the detected change in \( Z_{re} \) at \( T_A \) were initially defined as \( \Delta Z_{re} = (Z_{re})_{native} - (Z_{re})_{denatured} \), where \( (Z_{re})_{native} \) and \( (Z_{re})_{denatured} \) are the real part of sample impedance at the native state of spectrin (at a temperature 3°C less than \( T_A \)) and at the denatured state of spectrin (at a temperature 3°C greater than \( T_A \)), respectively. This change, however, levels off within a substantial temperature interval of about 6°C where it is superimposed on the continuous, thermally induced change in electrolyte conductivity. To compensate for the latter, which had no relation to the temperature-induced event at \( T_A \), the change in \( Z_{re} \), taking place over an equal temperature interval prior the denaturation at \( T_A \), was likewise determined and subtracted from the initial \( \Delta Z_{re} \). Using the same algorithm the amplitude of the real changes of \( Z_{im} \) at \( T_A \), \( \Delta Z_{im} \), was determined and corrected for the thermally induced change in electrolyte conductivity. Also, the corrected amplitudes of the changes in \( Z_{re} \) and \( Z_{im} \) at \( T_C \) were likewise determined and further used.

Based on the theory of dielectroscopy (Kell 1987; Klösgen et al. 2011) the frequency dependences of corrected amplitudes \( \Delta Z_{re} \) and \( \Delta Z_{im} \) were studied plotting the \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) for each denaturation temperature, \( T_A \) and \( T_C \). Complex plane (Nyquist) plot of \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) determined at the spectrin denaturation temperature, \( T_A \), had the form of a semicircle arc, situated below the argument axis (Fig. 6).

![Figure 6](image)

**Figure 6.** Complex plane plot of \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) at the denaturation temperature of spectrin (○) and of band 3 (●). \( \Delta Z_{re} \) and \( \Delta Z_{im} \) are the amplitudes of the changes of the real and imaginary parts of complex impedance, respectively, of packed Triton-X-100 shells at the indicated temperatures of denaturation. The critical frequencies of detected dielectric relaxations are indicated by \( f_{cr} \).

The complex plane plot of \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) determined at the band 3 denaturation temperature, \( T_C \), represented another semicircle arc, situated above the argument axis (Fig. 6). According to Kell (1987) and Klösgen et al. (2011) each semicircle arc in Fig. 6 revealed a dielectric relaxation of Debye type, i.e. polarisation mechanism with a single critical frequency, \( f_{cr} \) (a single relaxation time, \( \tau = 1/(2\pi f_{cr}) \)).

As shown in Fig. 6, the \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) plot at the spectrin denaturation temperature revealed only one dielectric relaxation due to the direct interaction of electric field with the dipole moments of spectrin (Ivanov et al. 2012; Ivanov and Paarvanova 2016). This finding is consistent with the basic conception that Triton-X-100 residues lacked closed lipid bilayer encapsulating them. The critical frequency, \( f_{cr} \), for this relaxation is implicitly determined by the top point of the corresponding semicircle arc, indicated by arrow in Figure 6. From multiple experiments (\( n = 5 \)) with shells, isolated and washed at 10 mM NaCl concentration this critical frequency was determined at 830 kHz (Fig. 6). With shells, isolated at 10 mM NaCl and washed by media with increased NaCl concentration up to 120 mM the critical frequency of spectrin relaxation linearly increased from 830 kHz to 9 MHz (not shown). The presence of glycerol, known for its ability to strongly increase the viscosity of solutions, linearly decreased the critical frequency of spectrin relaxation. For example, at 120 mM NaCl, this critical frequency was decreased from 9 MHz at zero % glycerol to about 5 MHz at 40% glycerol (not shown). These data are in full concert with the same dependences, recently reported for the spectrin skeleton of intact erythrocytes and impermeable EGMs (Ivanov and Paarvanova 2016) and indicate preserved molecular structure and dynamics of membrane skeleton in Triton-X-100 shells.

On the other hand the critical frequency, \( f_{cr} \), for the \( -\Delta Z_{im} \) vs. \( \Delta Z_{re} \) plot at the denaturation temperature of band 3 was determined 1300 kHz (\( n = 5 \)). It did not depend on the NaCl concentration (10–100 mM) and on the presence of glycerol (not shown). Both critical frequencies were not affected by the preliminary DIDS-treatment of erythrocytes.

Thermal dielectroscopy was employed to investigate the alterations of major proteins, spectrin and band 3, induced by preliminary freeze-thaw of Triton-X-100 shells. In Figure 7 these alterations are demonstrated by the temperature derivative of the resistance, \( Z_{re} \), measured at 100 kHz. A similar result was obtained using derivative thermograms of other components \( (Z_{im}, C_{re} \text{ and } C_{im}) \) of complex impedance, \( Z^* \), and capacitance, \( C^* \), measured at various frequencies between 100 kHz and 5 MHz (not shown). The above mentioned dielectric changes at \( T_A \) and \( T_C \) were altered in the residues subjected to preliminary freeze-thaw (Fig. 7, curves 2, 3 and 4), compared to control shells not subjected to subzero temperatures (Fig. 7, the curve 1). In addition, the frequency dependences of the amplitudes of respective
dielectric changes at \( T_A \) and \( T_C \) were used to determine the effect of freeze-thaw on respective dielectric relaxations. The effects produced by freeze-thaw on residues did not depend significantly on the value of freezing temperature (–6, –16 and –80°C) and on the duration of freezing (1 to 10 h), however, they strongly depended on the rate of thaw. When the frozen residues were rapidly thawed in a water bath at 20°C for 10 min the denaturation C peak reduced its height, broadened its half-width and preserved its area (Fig. 7, curve 2 compared to curve 1). Peaks A and C exhibit the heat denaturation of spectrin and anion exchanger, respectively.

![Figure 7. Effect of freeze-thaw on Triton-X-100 residues as revealed by thermal dielectroscopy. The \( dZ_{re}/dT \) is the temperature derivative of the resistance, \( Z_{re} \), measured at 100 kHz, of Triton-X-100 residues. Residues were prepared from EMs of intact erythrocytes (curves 1, 2 and 3) and from EMs of DIDS-treated erythrocytes (curve 4). The control residues (curve 1) were not frozen, while the frozen residues were subjected to rapid thaw (curve 2) and slow thaw (curves 3 and 4). Peaks A and C exhibit the heat denaturation of spectrin and anion exchanger, respectively.](image)

The damage produced by the freeze and slow thaw on the band 3 was partially prevented by the preliminary binding of DIDS to this protein. Compared to Triton-X-100 residues, obtained from intact EGMs and not subjected to subzero temperatures (Fig. 7, curve 1), the cooperativity and total area of the C peak were fully preserved in residues, obtained from DIDS-treated EGMs and subjected to freeze and slow thaw (Fig. 7, curve 4). In addition, the denaturation temperature, \( T_C \), of band 3 of latter residues was increased by 13°C (Fig. 7, curve 4), a shift characteristic to DIDS-treated EGMs (Snow et al. 1978). However, the protective effect of DIDS was partial as the relaxation frequency of band 3 in DIDS-treated residues was decreased from 1300 kHz in residues not subjected to freeze-thaw to 350 kHz in residues subjected to freeze-thaw (not shown). This outcome possibly indicates that freeze-thaw produced aggregation and clustering of band 3.

Compared to band 3, spectrin was mildly damaged by the freeze-thaw of Triton-X-100 residues. The freeze-thaw cycle did not affect the cooperativity of spectrin denaturation as expressed by the preserved height and half-width of the A peak. The inducing temperature, \( T_A \), of this peak was, however, slightly decreased (Fig. 7, curves 2 and 3 compared to curve 1). The relaxation frequency of dipoles associated to spectrin, 830 kHz at 10 mM NaCl, preserved its value after freeze and slow thaw of Triton-X-100 residues. This result also underlined the low cryosensitivity of spectrin, compared to that of band 3.

In contrast to the rapid thaw (Fig. 7, curve 2), the slow thaw of residues produced a shoulder of the A peak centered at a temperature 6°C above \( T_A \) (Fig. 7, indicated by arrow on the curves 3 and 4). The new denaturation was not affected by the preliminary stabilisation of band 3 with DIDS (Fig. 7, curve 4). The relaxation frequency of the protein involved was very low, about 200 kHz, possibly indicating the involvement of aggregated and clustered copies. This protein could be peripheral one, similar to spectrin, but its identity is not known at present.

**Discussion**

This study tested the structural stability of major proteins, spectrin and band 3, of Triton-X-100 shells of EGMs as represented by the temperature and cooperativity of their thermal denaturations. During the denaturations of spectrin at 49.5°C (\( T_A \)) and band 3 at 66°C (\( T_C \)) the capacitance, \( C_{re} \), and resistance, \( Z_{re} \), of the sample of Triton-X-100 shells both changed in the same direction, increasing at \( T_A \) and decreasing at \( T_C \) for any frequency within the 50 kHz–10 MHz interval. The changes in resistance appear consequent upon the changes in capacitance (dielectric polarisability) of Triton-X-100 shells and related changes in the resultant electric field. This conclusion is consistent with the above mentioned basic conception that
each Triton-X-100 shell represented a network of proteins devoid of closed lipid bilayer surrounding them and, in contrast to intact erythrocytes and their resealed EGMs (Ivanov and Paarvanova 2016) it was fully transparent to the incident electric field with any frequency from the indicated frequency interval.

As shown in Fig. 2, a great number of tiny spheres (sphere-shaped vesicles) were firmly attached to the spectrin skeleton of Triton-X-100 shells. These objects could be envisaged as liposomes (closed inside-out lipid bilayers) made up of the remainder of original lipid bilayers and band 3 copies of EGMs. We assume the attachment spots could include band 3 tetramers which avoided the solubilisation, preserved a portion of their original lipid milieu and remained strongly linked to spectrin via the ankyrin bridges of EGMs. This assumption is consistent to the result (Fig. 5) that within the entire frequency interval of 50 kHz–10 MHz the capacitance (dielectric polarizability) of Triton-X-100 shells decreased at $T_C$ indicating immobilization of dielectrically active segments of band 3 copies after the denaturation of band 3. This decrease in polarizability could be explained with the restriction imposed by the lipid bilayer of attached liposomes on the mobility of band 3 dielectrically active segments after band 3 denaturation at $T_C$.

By contrast, within the entire frequency interval of 50 kHz–10 MHz the capacitance (dielectric polarizability) of Triton-X-100 shells increased at $T_A$ (Fig. 5) indicating that dielectrically active segments of spectrin increased their mobility after the spectrin denaturation. Previous study (Ivanov and Paarvanova 2016) reported that the change of the capacitance (dielectric polarizability) of intact erythrocytes and impermeable EGMs at the spectrin denaturation temperature (49.5°C) depended on frequency; it decreased at low frequencies (50 kHz–3 MHz) and increased at higher frequencies (3–10 MHz). This frequency effect was explained by the adherence of a part of denatured spectrin to the encapsulating lipid bilayer resulting in partial immobilization of dielectrically active spectrin segments and hardening of erythrocyte plasma membrane. Combining the above two lines of evidence we arrive again at the basic conception that Triton-X-100 shells lacked an encapsulating lipid bilayer thus the dielectrically active segments of spectrin were free to demonstrate higher mobility in the denatured than in native state of spectrin.

Fig. 7 shows a damage of the band 3 protein as well as the spectrin–actin membrane skeleton after freezing and thawing the Triton-X-100 residues. The freeze-thaw of residues had minor effect on spectrin denaturation at 49°C although an additional protein denaturation appeared at 55°C. The freeze and rapid thaw of Triton-X-100 residues resulted in a strong reduction of cooperativity of band 3 denaturation while the slow thaw completely eliminated the peak of this denaturation. The latter outcome is indicative of a substantial denaturation and/or detachment of band 3 copies of the residues. These effects of freeze-thaw were prevented in residues obtained from DIDS-treated EGMs supporting the view that the main damage implicated the membrane protein band 3. Additional study is needed to elucidate the possible damage induced by freeze/thaw on the band 3 linkage to membrane skeleton. In general, these findings are in line with other reports (Woelders and Malva 1998) that slow thaw is particularly damaging to cells during their recovery after cryopreservation. The above results could help design new experiments for finding optimal conditions to reduce the freeze damage to erythrocytes and other cells. Based on the similarity of the general structure of plasma membrane in erythrocytes and nucleated animal cells these results could be helpful for cryobiology, cryosurgery and the cryopreservation of cells and tissues.

Compared to deoxyribonucleic acids the sensitivity of spectrin denaturation temperature to formamide was several times higher (Ivanov 2001; Paarvanova et al. 2012). This outcome could be exploited for the design of drugs, which could be targeted to the major peripheral proteins, homologous to erythrocyte spectrin, in human and animal cells.

Conclusions

Heat and the freeze greatly affected the structural stability and dielectric properties of erythrocyte membrane skeletons, released after mild extraction of erythrocyte ghost membranes by Triton-X-100. The thermal methods applied well differentiated the thermal (structural) stability of spectrin and band 3, and the contributions of these major proteins to the dielectric properties of Triton-X-100 shells. The results presented indicate that Triton-X-100 shells preserved the native structure of spectrin and band 3. The Triton-X-100 shells appear as a helpful model for exploration on the structure and dynamics of erythrocyte plasma membrane proteins and their change under a variety of adverse conditions, especially during freeze and slow/rapid thaw.

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